## Assignment-1<sup>1</sup> Definitions, Energy band Diagrams and Ohms Law

Prerequisite for solving the below problems is thorough understanding of Lecture notes-1. All the constants are given in table-1 of the lecture notes-1. Any missing data may be suitably assumed and stated.

$$n = \frac{dv}{AM} = \frac{A_0 dv \times 10^3}{A}$$

where d= density,  $kg/m^3$ ; v=valence, free electrons per atom; A= atomic weight; M= weight of atom of unit weight, kg;  $A_0$ =Avogadro's number, molecules/mole.

- 1. For the ground state (n = 1) of the hydrogen atom, show that the radius is 0.53 Å.
- 2. A conducting line on an IC chip is 2.8 mm long and has a rectangular cross section  $1 \times 4 \ \mu m$ . A current of 5mA produces a voltage drop of 100mV across the line. Determine the electron concentration given that the mobility is  $500 \ cm^2/V s$ .
- 3. A flat aluminum strip has a resistivity of  $3.44 \times 10^{-8} (\Omega m)$ , a cross-sectional area of  $2 \times 10^{-4} mm^2$ , and a length of 5 mm. What is the voltage drop across the strip for a current of 50 mA.
- 4. For the aluminum strip described above in Prob.2, what current exists if the voltage across the strip is  $30\mu V$ ?
- 5. The specific density of tungsten is  $18.8g/cm^3$ , and its atomic weight is 184.0. Assume that there are two free electrons per atom. Calculate the concentration of free electrons. HINT: Use the equation mentioned above.
- 6. (a) Compute the conductivity of copper for  $\mu = 34.8 \ cm^2/V s$  and density  $d=8.9 \ g/cm^3$ .
  - (b) If an electric field is applied across such a copper bar with an intensity of 10 V/cm, find the average velocity of the free electrons.
- 7. Compute the mobility of the free electrons in aluminum for which the density is  $2.7 \ g/cm^3$  and the resistivity is  $3.44 \times 10^{-6} \ \Omega cm$ . Assume that aluminum has three valence electrons per atom.
- 8. Calculate the Bohr radius (Å) and energy (eV) for n=1,2,3.

<sup>&</sup>lt;sup>1</sup>Due date: 27th August 2006, 3:30 pm. To be submitted in the Office.